



Standard Test Method for pH of Activated Carbon¹

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1. Scope

1.1 This test method covers determination of the pH of a water extract of activated carbon.

1.2 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see Section 6.

2. Referenced Documents

2.1 ASTM Standards:

D 1193 Specification for Reagent Water²

D 1293 Test Methods for pH of Water²

D 2867 Test Method for Moisture in Activated Carbon³

E 300 Practice for Sampling Industrial Chemicals⁴

3. Summary of Test Method

3.1 An activated carbon sample is boiled in reagent water using a reflux condenser to recycle water vapor. The particles of carbon are filtered out, the filtrate cooled to 50°C and the pH of the filtrate determined by electrometric measurement.

4. Significance and Use

4.1 When a fluid containing an adsorbate is passed through a bed of activated carbon, chemical reactions may take place between the activated carbon, its other noncarbonaceous constituents, and the adsorbate containing fluid. The pH of the carbon may be a significant parameter of such a reaction and therefore may be an important characteristic of the carbon.

5. Apparatus and Materials

5.1 *Analytical Balance*, capacity 100 g, precision ± 0.01 g.

5.2 *Hot Plate*.

5.3 *Glassware for Boiler-Reflux Condenser Apparatus* (Fig. 1)—Items shown are for guidance only, providing a convenient set of equipment available off-the-shelf from many laboratory supply houses. The “all-glass” elements, with standard-taper

¹ This test method is under the jurisdiction of ASTM Committee D-28 on Activated Carbon and is the direct responsibility of Subcommittee D28.04 on Gas Phase Evaluation Tests.

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² *Annual Book of ASTM Standards*, Vol 11.01.

³ *Annual Book of ASTM Standards*, Vol 15.01.

⁴ *Annual Book of ASTM Standards*, Vol 15.05.

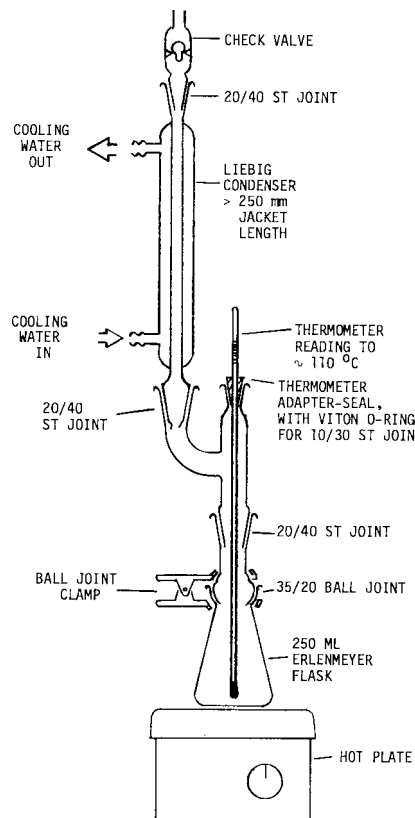


FIG. 1 Boiler-Reflux Condenser Apparatus

and ball joints, provide freedom from contamination and maintenance. A check valve in the position shown is essential to relieve pressure buildup while minimizing loss of vapor.

5.4 *Thermometer*, glass, approximately 0 to 120°C, long enough to be read at 100°C when inserted to bottom of Erlenmeyer flask in Fig. 1.

5.5 *Thermometer*, glass, approximately 20 to 55°C.

5.6 *Graduated Cylinder*, 100-mL.

5.7 *Beaker*, 200-mL.

5.8 *Filter Funnel*.

5.9 *Filter Paper*, qualitative, medium flow rate, 12.5 cm or larger in diameter.

5.10 *Timer*.

5.11 *pH Meter*, in accordance with Test Methods D 1293, Type II (automatic thermal compensation preferred).

5.12 *Reagent Water*, in accordance with Specification D 1193, Type II.

6. Hazards

6.1 The test method involves transfer of boiling water between containers; appropriate tongs or gloves should be used. In addition, the use of an electric hot plate and pH meter (if line-powered) poses a shock hazard. This equipment must be grounded and insulated in accordance with UL standards and electrical codes.

7. Sampling

7.1 Guidance in sampling activated carbon is given in Practice E 300.

8. Calibration and Standardization

8.1 Use the procedure of Test Methods D 1293 to standardize the pH meter.

9. Procedure

9.1 Determine the moisture content of the carbon in accordance with Test Method D 2867. Calculate the weight of moist carbon equivalent to 10 g on a dry basis.

9.2 Weigh out a sample of carbon equivalent to 10.00 ± 0.01 g on a dry basis. Remove boiler flask from apparatus and add carbon sample.

9.3 Bring approximately 110 mL of reagent water to a gentle boil. Measure 100.0 ± 0.1 mL in graduated cylinder while water is hot. Immediately add this water to the carbon in the flask. Reconnect joints to flask seated on hot plate.

9.4 Bring water in flask to a boil, using thermometer reading to assure that no false boiling, due to gases trapped in carbon, occurs.

9.5 Boil gently for 900 ± 10 s.

9.6 Remove the flask from the hot plate and filter its contents immediately through the filter paper premoistened with the distilled water used for the test. Catch the filtrate in a 500-mL vacuum filter flask, being careful to prevent carbon-fines from overrunning into the flask.

9.7 Cool the filtrate to $50 \pm 5^\circ\text{C}$.

9.8 Measure pH as described in Test Methods D 1293.

10. Calculation

10.1 If the pH meter is calibrated in pH units, read value from pH meter. If the meter reading is in voltage units, use the equation from the Terminology section of Test Methods D 1293.

11. Report

11.1 Report the following information:

11.1.1 Source of sample,

11.1.2 Type or designation of activated carbon,

11.1.3 Supplier name,

11.1.4 Supplier grade designation,

11.1.5 Supplier lot and batch number,

11.1.6 Moisture content in accordance with Test Method D 2867,

11.1.7 pH to the nearest 0.1 unit,

11.1.8 Temperature of measurement to the nearest 1°C ,

11.1.9 Date of test,

11.1.10 Name and signature of technician performing test, and

11.1.11 Name and signature of supervisor approving test.

12. Precision and Bias

12.1 An interlaboratory comparison of the pH of four types of activated carbons using essentially the above procedure was carried out by four laboratories. The results were as follows:

Sample	Number of Tests	Average pH	Standard Deviation
2N	11	3.65	0.18
11N	12	3.64	0.19
12N	11	3.56	0.14
1D	11	5.57	0.67
4D	12	5.62	0.60
9D	12	5.65	0.60
3B	11	8.82	0.38
5B	12	8.93	0.30
10B	12	8.90	0.29
6P	11	7.91	0.17
7P	12	8.07	0.41
8P	12	7.85	0.23

12.2 The measured pH values are in agreement with the activated carbon types (N, wood-base; D, lignite; B and P, bone-char). The standard deviations also appear in agreement with other work, in that they are higher for neutral conditions (near pH 7) than for strongly acidic or basic conditions. Some of the observed variation is in the samples themselves. A standard deviation of 0.6 for pH values near 7 and 0.3 for values at the ends of the pH scale appears readily achievable by this test method.

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